

Determination and Evaluation of Heavy Metal Pollution in Greenhouse Regions of Antalya (Turkey)

Bülent Topcuoğlu

Akdeniz University Vocational School of Technical Sciences, 07058 Antalya TURKEY

Phone: 0090242-3106751; fax: 0090242-2274785

E-mail: btoglu@ akdeniz.edu.tr

Abstract: *To evaluate the heavy metal pollution of intensive greenhouse regions of Antalya, the relations among the soil, greenhouse plants and groundwater properties and heavy metal characteristics were determined. Additional to routine water and soil analysis, a sequential extraction procedure was used to estimate the availability of heavy metals (Zn, Cd, Ni and Pb) in greenhouse soils and several environmental pollution indexes were used to evaluate the size of pollution and risks.*

Groundwaters of greenhouse area have low conductivity but high nitrate content. Heavy metal contents of groundwaters were below the permissible levels. Average total Cd, Ni, Pb and As contents of groundwaters were below the pollutant limits, but in sampling sites for Zn, Ni, Pb and As concentrations were exceeded permissible limits. However, the average heavy metal evaluation index (HEI) values for all metals in groundwaters were below the critical value.

The concentration of heavy metals with the exception of Ni in soils of greenhouses was generally below the referenced limits. Soil metal speciation showed that the greatest percentage of all metals was present in the residual form, and the mobility of metals declined in the following order: As>Cd>Zn>Pb>Ni. Single factor and composite pollution coefficient values of all metals with the exception of Ni were not exceeded critical limit. Anthropogenic and enrichment factor indexes of greenhouse soils indicate that both two parameter showed similar trends and 1 to 5 fold metal enrichment by anthropogenic inputs compared to uncontaminated soil. Potential ecological risk indexes of soil metals except Ni were found below the threshold value that indicate these metals have a low risk to surrounding environment.

All heavy metal concentrations with the exception of Cd in tomato fruits were found below the permissible heavy metal limits. Although soil Ni concentration was exceeded the pollutant limits, possibly due to low soil mobility factor of Ni, concentration of Ni in tomato fruit was found very low. However, target hazard quotient (THQ) of tomato fruits was found below the critical value and thus it can be presumable no health risk for Cd metal in short or medium terms.

In a comprehensive manner, in addition to total concentrations referenced, environmental risk evaluation methods and soil speciation studies may provide useful information for assessing metal bioavailability and environmental risk

Keywords: *Greenhouse Soils, Groundwater, Tomato, Metal Speciation, Metal Mobility, Pollution Indexes*

1. Introduction

Due to intensive use of agrochemicals in greenhouse soils, heavy metals are become to common pollutants in greenhouse soils and adjacent environment. Repeated amendements of organic matter and intensive use of fertilizers, metal-enriched chemicals and biocides may cause soil and environmental pollution in greenhouses. Although greenhouse areas a have great impact on environment due to intensive use of agrochemicals, little attention has been paid to metal accumulation in greenhouse plants, metal contents of ground waters around greenhouses and heavy metal speciation and metal bioavailability and environmental pollution assessment in greenhouse soils with respect to comprehensive and integrated environmental evaluation.

stored in polyethylene bags sealed awaiting analysis.

Electrical conductivity (EC) and pH were measured a soil:water ratio of 1:2. cation exchange capacity (CEC) was determined by 0.1 M NH_4Ac extraction; CaCO_3 content was determined by the calcimeter; organic carbon was measured by wet oxidation; and texture was determined by Bouyoucos hydrometer method.

Sequential extraction method [4] was applied to soil samples to identify metal fractions.

For the determination of ‘total’ heavy metal concentrations, soil samples were digested in aqua regia (1:3 HNO_3/HCl) and HClO_4 according to the international standard [5] Zn, Cd, Ni and Pb concentrations of water and greenhouse soil samples were analysed using ICP-MS under optimised measurement conditions, and values were adjusted for oven dried (12 h at 105 °C) material.

Selected environmental pollution indexes for water samples ‘Heavy Metal Evaluation Index (HEI) of the Groundwaters’ [6], as for soil samples ‘Mobility of Metals’ [7], ‘Single-Factor and Composite Pollution Index of Soils’ [8], Anthropogenic Factor (AF) [9] and ‘Enrichment Factor (EF) Indexes of Soil’ [10], ‘Potential Ecological Risk Factor Indexes’ [11], and as for plant samples ‘Heavy Metal Transfer (Bioconcentration) Factor’ and ‘Target Hazard Quotient (THQ) of Food’ [12] were used for comprehensive and integrated evaluation of parameters.

Statistical analyses were performed by using SPSS-16 for Windows program

3. Result and Discussion

3.1 Ground Water Properties

Certain groundwater characteristics and total heavy metal contents of greenhouse areas are shown in Table 1. Groundwaters in greenhouse areas have generally slightly alkaline reaction, low electrical conductivity. Total nitrate content of groundwater has exceeded maximum permissible limits for drinking waters. High concentration of nitrate is of course may be due to highly intensive agricultural practices for all season. Average total Cd, Ni, Pb and As contents were below the permissible pollution limits, but in some sampling sites permissible limits for Zn, Ni, Pb and As were exceeded.

TABLE I: The analytical characteristics and heavy metal concentrations of the groundwaters

| Site | pH | EC, $\mu\text{S cm}^{-1}$ | NO_3 , mg L^{-1} | Zn, $\mu\text{g L}^{-1}$ | Cd, $\mu\text{g L}^{-1}$ | Ni, $\mu\text{g L}^{-1}$ | Pb, $\mu\text{g L}^{-1}$ | As, $\mu\text{g L}^{-1}$ |
|------------|------|---------------------------|------------------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|
| 1 | 7,52 | 479 | 20,0 | 268 | 2,41 | 4,02 | 52,9 | 8,7 |
| 2 | 7,58 | 486 | 21,5 | 131 | 0,08 | 21,10 | 3,3 | 3,8 |
| 3 | 7,53 | 492 | 18,0 | 71 | 0,16 | 3,68 | 3,6 | 10,4 |
| 4 | 7,35 | 569 | 22,7 | 97 | 0,06 | 4,64 | 1,2 | 9,3 |
| 5 | 7,43 | 640 | 17,5 | 106 | 0,10 | 5,21 | 4,8 | 20,4 |
| 6 | 7,35 | 655 | 32,8 | 101 | 0,07 | 4,52 | 2,2 | 13,9 |
| 7 | 7,16 | 793 | 37,2 | 105 | 0,40 | 4,44 | 5,5 | 20,5 |
| 8 | 7,27 | 777 | 44,0 | 83 | 0,05 | 2,85 | 4,1 | 17,5 |
| Mean | 7,39 | 611 | 26,7 | 120 | 0,41 | 6,30 | 9,7 | 13,1 |
| St.D. | 0,14 | 127 | 9,9 | 62,2 | 0,81 | 6,01 | 17,5 | 6,0 |
| Limits[12] | | | 10 | 200 | 3 | 20 | 10 | 20 |

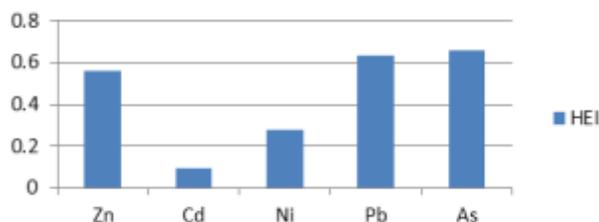


Fig. 2: Heavy metal evaluation index (HEI) of the groundwaters

The mean heavy metal evaluation index (HEI) values for all of metals in groundwaters were below the critical value 40. Thus, all of groundwater samples may be considered as less contaminated and may be acceptable clean.

3.2 Soil Properties

Certain soil characteristics and heavy metal contents of greenhouse areas are shown in Table 2 and Table 3. These greenhouse soils have generally slightly alkaline reaction, moderate CEC, low EC values and highly calcareous. These soil characteristics, together with irrigation by sprinklers and agricultural practices, suggest that intensive greenhouses agriculture is the main cause of soil contamination by heavy metals and that theoretically the heavy metal availability will be lower [13].

The total metal contents of the experimental soil and their pollutant limits were given in Table 3. The results of Table 3 ranged ($\mu\text{g g}^{-1}$) from 57 to 249 for zinc with a mean of 138; 0.23 to 0.78 for cadmium with a mean of 0.44; 56 to 215 for nickel with a mean of 122; 9,3 to 30,4 for lead with a mean of 19,4; and 4 to 34,3 for arsenic with a mean of 11,9. Average total metal contents except zinc were above the limits of european union, 86/278/EEC [14] directive to agricultural soils with $\text{pH}>7$. Nickel concentration in all soil samples were higher than limit values. According to these data, the order for the average content of metals in analysed samples is $\text{Ni}>\text{Zn}>\text{Pb}>\text{As}>\text{Cd}$.

TABLE II: The analytical characteristics of soils

| Site | CaCO ₃ , % | pH (H ₂ O) | EC, micS cm ⁻¹ | CEC, meq ⁻¹⁰⁰ g | Org. C., g ^{-kg} | Clay, % |
|-------|-----------------------|-----------------------|---------------------------|----------------------------|---------------------------|---------|
| 1 | 15,0 | 7,85 | 574 | 15,25 | 1,28 | 10,05 |
| 2 | 17,2 | 7,49 | 892 | 28,80 | 2,64 | 8,20 |
| 3 | 40,8 | 7,57 | 2020 | 19,40 | 3,53 | 11,02 |
| 4 | 20,9 | 7,57 | 1444 | 22,47 | 3,17 | 11,01 |
| 5 | 19,5 | 7,56 | 1914 | 31,85 | 4,20 | 9,76 |
| 6 | 25,1 | 7,80 | 1076 | 23,98 | 3,19 | 8,76 |
| 7 | 5,7 | 7,50 | 1535 | 17,80 | 2,59 | 9,39 |
| 8 | 35,9 | 7,96 | 374 | 13,80 | 1,52 | 8,88 |
| Mean | 22,51 | 7,66 | 1229 | 21,69 | 2,77 | 9,63 |
| St.D. | 1,13 | 0,18 | 601 | 6,36 | 0,98 | 1,03 |

TABLE III: Total metal contents ($\mu\text{g g}^{-1}$ dry wt) of the greenhouse soils and their pollutant limits

| Site | Zn | Cd | Ni | Pb | As |
|-------------|--------|----------|------|------|------|
| 1 | 173 | 0,50 | 118 | 12,3 | 6,5 |
| 2 | 249 | 0,35 | 215 | 13,7 | 6,6 |
| 3 | 170 | 0,23 | 90 | 9,3 | 8,7 |
| 4 | 126 | 0,45 | 130 | 20,0 | 7,5 |
| 5 | 112 | 0,62 | 129 | 25,1 | 34,3 |
| 6 | 88 | 0,78 | 56 | 30,4 | 17,9 |
| 7 | 127 | 0,33 | 64 | 28,6 | 4,0 |
| 8 | 57 | 0,25 | 175 | 15,9 | 9,9 |
| Mean | 138 | 0,44 | 122 | 19,4 | 11,9 |
| St.D. | 5,9 | 0,19 | 5,39 | 7,8 | 9,9 |
| Limits [14] | 20-300 | 0.03-0.3 | 50 | 2-20 | 1-7 |

Metal Speciation

Concentrations of Zn, Cd, Ni, Pb and As in soil fractions were given in Figure 3. Irrespective of sampling point, the distribution of metals in greenhouse soil samples generally followed the order below for the metals studied.

Zn: F1<F3<F4<F2<F5

Cd: F2<F3<F4<F1<F5

Ni: F3<F1<F2<F4<F5

Pb: F3<F2<F1<F4<F5

As: F3<F4<F2<F1<F5

The study of the distribution of metals showed that the greatest percentage of all metals was present in the residual fraction (F5). However, F1 and F2 fractions of Zn; Cd and As metals were higher than other metals. This property possibly gives these metals a high mobility. The most mobile metal fraction was detected in As and the most immobile metal fraction was detected in Ni. Ni largely (97,6 %) associated with residual phase. The residual phase represents metals largely embedded in the crystal lattice of the soil fraction and should not be available for remobilization except under very harsh conditions [7].

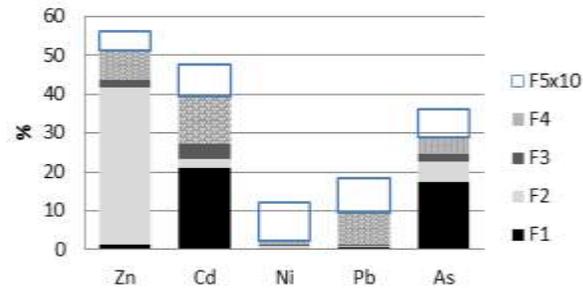


Fig. 3: Concentrations of Zn, Cd, Ni, Pb and As in soil fractions (F5 values are higher 10 fold than given values)

Mobility of metals

Due to some metal forms are strongly bound to soil components than those extracted in F1, F3 and F3, the mobility of metals in soil samples may be evaluated on the basis of absolute and relative content of fractions weakly bound to soil component. Relative index of metal mobility was calculated as a 'mobility factor' (MF) [15] on the basis of the following equation:

$$MF : ((F_1+F_2+F_3))/((F_1+F_2+F_3+F_4+F_5)) \times 100$$

This equation largely describes the potential mobility of metals. The MF values were considerably higher for As, Zn and Cd. The high MF values have been interpreted as symptoms of relatively high lability and biological availability of heavy metals in soils [15]. The results of the present study suggest that the mobility of the metals declines in the following order: As>Cd>Zn>Pb>Ni (Figure 4).

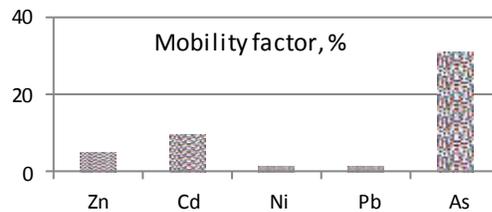


Fig. 4: Average metal mobility of greenhouse soils

Contamination Evaluation of heavy metals

Anthropogenic factor (AF) and Enrichment factor (EF) indexes of soil

Anthropogenic and enrichment factor indexes of greenhouse soil were given in Figure 5. Estimated values of AF for the heavy metals determined in the soil samples with respect to the uncontaminated soil in the same area were generally greater than one and range from 1,48-0,81 (Zn), 2,70-1,64 (Cd), 2,56-1,38 (Ni), 2,62-1,41 (Pb) and 5,53-3,00 (As). This indicates a, 1 to 5 fold metal enrichment by anthropogenic inputs compared to uncontaminated soil.

Enrichment factor values showed similar trends with anthropogenic factor values. Mean Enrichment factor Zn was below the moderate level and approximate to depletion level.

Although total Ni concentration of greenhouse soils were above typical soil concentrations and permissible contaminant limits, enrichment factor is very low and also in some sampling sites enrichment factor of Ni were in depletion level. This may be inferred that Ni abundance of parent material of soil is very high and there are less Ni contaminant sources.

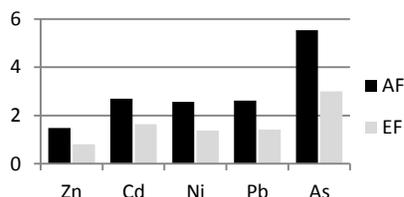


Fig. 5: Anthropogenic (AF) and enrichment factor (EF) indexes of the greenhouse soils

Single-factor (Pi), index composite pollution (PN) index and potential ecological risk factor index (Er)

Single-factor and composite pollution indexes and potential ecological risk indexes of heavy metals in greenhouse soils are summarized in Figure 6. It is clear that all contamination coefficients with the exception of Ni were not exceeded critical value 1. Contamination coefficient of Ni was exceeded critical value in all sampling sites. Although contamination coefficient of other metals was low, due to higher coefficient value of Ni, composite pollution index was determined in heavy pollution risk group.

The average monomial risk factors, Er of heavy metals in greenhouse soils were ranked in the following order Zn<Pb<As<Cd<Ni. The average monomial risk for heavy metals was found below the 40 that indicate all metals posed low risk to surrounding ecosystem. In order to quantify the overall potential ecological risk of observed metals in the greenhouse soils, general ecological risk factor (RI) value was calculated as the sum of all the risk factors. Average RI value were found 19, 17 and below the ecological risk level.

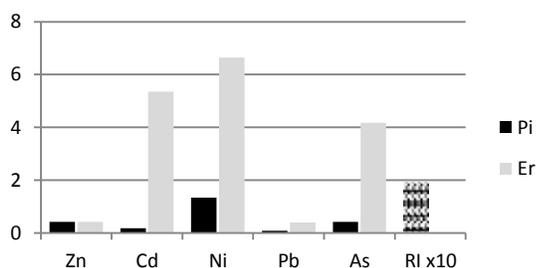


Fig. 6: Single factor index (Pi) of each metal, composite pollution index (PN) and and potential ecological risk indexes (Er) of metals

3.3 Plant Properties

Plant heavy metal content

Heavy metal concentration of fresh tomato fruits are shown in Figure 7.

All heavy metal concentrations with the exception of Cd were found below the permissible heavy metal limits for fresh vegetables [16]. Average Cd concentrations in tomato fruit were exceeded referenced limit (0,02 mgkg⁻¹) [16]. Although soil Ni concentration was exceeded the pollutant limits, possibly due to low soil mobility factor of Ni, concentration of Ni in tomato fruit was found very low.

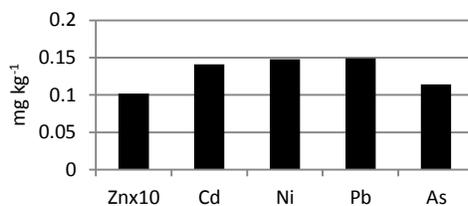


Fig 7: Heavy metal contents of tomato fruits grown in greenhouses (Zn element value is higher 10 fold than given values)

Heavy metal transfer factor (TF) and Target Hazard Quotient (THQ) of Tomato Fruit

TF and THQ values are presented in Figure 8. The trend of TF value ranges were: Cd>Zn>Pb>Ni>As. The highest average TF was found 3,58 for Cd in tomato fruits. The second high TF was found for Zn in the either tissues of tomato plant. These possibly might be due to higher mobility factor of Cd and Zn in the greenhouse soil (Figure 4) and may be due to soluble metal participations by agricultural practices or antropogenic factors. The mobility of metals from soil to plants is a function of the physical and chemical properties of the soil and of plant species, and is altered by innumerable environmental and antropogenic factors [17].

High Cd accumulation in tomato fruits may be possibly caused by high metal mobility and high enrichment factors of soil Cd. However, although As has the most mobile metal in greenhouse soils, bioconcentration factor was low.

As can be seen mean THQ values were found below the critical value 1 with the exception of Cd metal in sampling site 8 resulted as 1,13 THQ ratio. According to these results there can not be proposed a health risk for Cd metal in short or medium term.

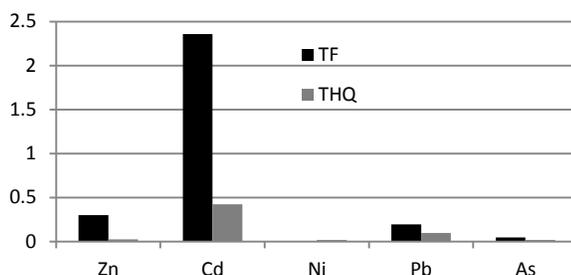


Fig 8: Heavy metal transfer factor and Target Hazard Quotient (THQ) of tomato

4. Conclusions

According to water analysis, groundwaters were not saline in nature but nitrate contents were exceeded maximum permissible limits for drinking waters. Average total Cd, Ni, Pb and As contents were below the permissible pollution limits, but in some sampling sites permissible limits for Zn, Ni, Pb and As were exceeded. However, the average heavy metal evaluation index (HEI) values for all metals in groundwaters were below the critical value. Thus generally, it can be concluded that all ground waters in regional size may be considered less contaminated, and in point of heavy metals and pollutants is in acceptable limits. High nitrate contents of groundwaters due to agricultural activities all season seem the main threats for public health.

The concentration of heavy metals with the exception of Ni in soils of Antalya greenhouses were generally below the limits referenced by the 86/278/EEC directive to agricultural soils with pH >7. Ni concentrations in all soil sample were higher than limit values. Soil metal speciation showed that the greatest percentage of all metals was present in the residual form, and the mobility of metals declined in the following order: As>Cd>Zn>Pb>Ni. Thus, although Ni was the most important threatening metal as total concentration, its mobility was found very low. Single factor and composite pollution coefficient values of all metals with the exception of Ni were not exceeded critical limit. However, due to high total concentration of Ni in greenhouse soils, composite pollution index was determined in heavy pollution risk group. Anthropogenic and enrichment factor indexes of greenhouse soils indicate that both two parameter showed similar trends and 1 to 5 fold metal enrichment by anthropogenic inputs compared to uncontaminated soil. Mean Enrichment factor Zn was below the moderate level and approximate to depletion level.

Potential ecological risk indexes of soil metals except Ni were found below the threshold value that indicate these metals have a low risk to surrounding environment.

All heavy metal concentrations with the exception of Cd in tomato fruits were found below the permissible heavy metal limits. Although soil Ni concentration was exceeded the pollutant limits, possibly due to low soil mobility factor of Ni, concentration of Ni in tomato fruit was found very low. High Cd accumulation in tomato fruits may be possibly caused by high metal mobility and high enrichment factors of soil Cd. However, according to target hazard quotient value of tomato fruit, it was found that THQ of tomato fruits is below the critical value and thus it can be presumable no health risk for Cd metal in short or medium terms.

As it is seen, the comparison results of risk values of heavy metals based on different methods show that there are several disagreements. Most of these paradoxes in evaluation are mainly depend on the total content of heavy metals as a criterion for determining their potential effect on the environments. Whereas in a comprehensive manner, in addition to total concentrations, environmental risk evaluation methods and soil speciation studies will provide useful information for assessing metal bioavailability or toxicity.

5. Acknowledgements

This research was sponsored by TUBITAK (The Scientific and Technological Council of Turkey). Author would like to thank to TUBITAK for the financial support of the project (TOVAG-1110711).

6. References

- [1] Cruz, J.V., Silva, M.O., Diaz, M.I., Prudencio, M.I., 2013. Groundwater composition and pollution due to agricultural practices at sete cidades volcano (Azores, Portugal). *Applied Geochemistry*, 29:162-173.
<http://dx.doi.org/10.1016/j.apgeochem.2012.11.009>
- [2] Heaton, T., Stuart, M., Sapiano, M., Sultana, M., 2012. An isotope study of the sources of nitrate in Malta's groundwater. *J. Hydrology*, 414(415):244-254.
<http://dx.doi.org/10.1016/j.jhydrol.2011.10.037>
- [3] Sundaray, S.K., Nayak, B.B., Lin, S., Bhatt, D., 2011. Geochemical speciation and risk assessment of heavy metals in the river estuarine sediments-A case study: Mahanadi basin, India[J]. *Journal of Hazardous Materials*, 186:1837-1846.
<http://dx.doi.org/10.1016/j.jhazmat.2010.12.081>
- [4] Tessier, A., Campbell, P.G.C., Bison, M., 1979. Sequential extraction procedure for the speciation of particulate trace metals. *Anal. Chem.* 51, 844-851.
<http://dx.doi.org/10.1021/ac50043a017>
- [5] ISO 11446 International Standard, 1995. Soil quality-extraction of trace elements soluble in aqua regia. 03-01.
- [6] Edet, A.E., Offiong, O.E., 2002. Evaluation of water quality pollution indices for heavy metal contamination monitoring. A study case from Akpabuyo-Odukpani area, Lower Cross River Basin (southeastern Nigeria), *Geojournal*, 57, 295-304.
<http://dx.doi.org/10.1023/B:GEJO.0000007250.92458.de>
- [7] Yusuf, K.A., 2007. Sequential extraction of lead, copper, cadmium and zinc in soils near ojota waste site. *Journal of Agronomy* 6(2):331-337.
<http://dx.doi.org/10.3923/ja.2007.331.337>
- [8] Cheng, J.L., Shi, Z., Zhu, Y.W., 2007. Assessment and mapping of environmental quality in agricultural soils of Zhejiang province, China. *Journal of Environmental Sciences*, 19:50-54.
[http://dx.doi.org/10.1016/S1001-0742\(07\)60008-4](http://dx.doi.org/10.1016/S1001-0742(07)60008-4)
- [9] Adamu, C., I., Nganje, T.N., 2010. Heavy metal contamination of surface soil in relationship to land use patterns: A case study of Benue state, Nigeria. *Materials Science and Applications*, 1:127-134.
<http://dx.doi.org/10.4236/msa.2010.13021>
- [10] Reimann, C., de Caritat, P., 2005. Distinguishing between natural and anthropogenic sources for elements in the environment: Regional geochemical surveys versus enrichment factors. *The science of the total environment*, 337:91-107.
<http://dx.doi.org/10.1016/j.scitotenv.2004.06.011>
- [11] Hakanson, L., 1980. An ecological risk index for aquatic pollution control: A sedimentological approach. *Water Research*, 14: 975-1001.
[http://dx.doi.org/10.1016/0043-1354\(80\)90143-8](http://dx.doi.org/10.1016/0043-1354(80)90143-8)
- [12] United States, Environmental Pollution Agency (USEPA), 2007. Integrated risk information system. Available from: (<http://cfpub.epa.gov/ncea/iris/index.cfm?fuseaction=iris.showSubstanceList>).
- [13] Gil, C., Boluda, R., Ramos, J. 2004. Determination and evaluation of cadmium, lead, and nickel in greenhouse soils of Almeria (Spain). *Chemosphere*, 55, 1027-1034.
<http://dx.doi.org/10.1016/j.chemosphere.2004.01.013>
- [14] C.E.C. (Council of the European Communities) 1986. Directive of 12 June 1986 on the protection of the environment, and in particular of the soil, when SS is used in agriculture (86/278/CEE). *Official Journal of the European Communities*, L181, 6-12.
- [15] Soon, Y.K., Abboud, S., 1990. Trace elements in agricultural soils of North-western Alberta. *Can.J. Soil Sci.* 70, 277-288.
<http://dx.doi.org/10.4141/cjss90-029>

- [16] WHO/FAO, 2007. Joint FAO/WHO food standart programme codex alimentarius commission 13th session. Report of the thirty-eight session of the codex committee on food hygiene, Houston, USA.
- [17] Zurera, G., Estrada, B., Rincon, F., Pozo, R. 1987. Lead and cadmium contamination levels in edible vegetables. Bull. Environ. Cont. Toxicol., 38:805-812
<http://dx.doi.org/10.1007/BF01616705>